**CHEM 2411: Organic Chemistry (Part 1)**

**Fall 2019; Section 02; TR 9:30 AM – 10:45 AM; TLC 1200**

**Instructor:** Dr. Partha Ray (Office: TLC 2123; Phone: (678) 839-6023; email: psray@westga.edu)


**Course Objective:** The goal of this course is for you to learn the fundamental principles, rules, and concepts of Organic Chemistry and use them to solve problems. You will utilize and build on these in CHEM 3422 (Organic Chemistry, part 2).

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<thead>
<tr>
<th>Test</th>
<th>Date</th>
<th>Chapters/Topics</th>
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<tbody>
<tr>
<td>1</td>
<td>Sept 5</td>
<td>Sections not covered in class are shown in parentheses</td>
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<td></td>
<td>1: Structure and Bonding</td>
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<td>2: Polar Covalent Bonds; Acids &amp; Bases (7)</td>
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<td>3: Alkanes and Their Stereochemistry</td>
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<td>4: Cycloalkanes and Their Stereochemistry</td>
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<td>2</td>
<td>Oct 1</td>
<td>5: Stereochemistry at Tetrahedral Centers (4,11,12).</td>
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<td>6: An Overview of Organic Reactions (7,8,9,11)</td>
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<td>7: Alkenes: Structure and Reactivity (1,10)</td>
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<td>3</td>
<td>Oct 29</td>
<td>8: Alkenes: Reactions and Synthesis</td>
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<td>9: Alkynes: An Introduction to Organic Synthesis</td>
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<td>10: Organohalides</td>
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<td>4</td>
<td>Nov 21</td>
<td>11: Reactions of Alkyl Halides: Nucleophilic Substitutions and Eliminations</td>
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<td>12: Structure Determination: Mass Spectrometry and Infrared Spectroscopy (3,4,5,8)</td>
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<td>13: Nuclear Magnetic Resonance Spectroscopy (4,12)</td>
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Dec 3 & 5  

Review for final

**Thurs Dec 12**  
ACS Final Exam: 8:00 – 10:00 AM

**Grades:** A: 85-100; B: 75-84; C: 60-74; D: 50-59; F: 0-49

**Final Grade:** The final grade will be based on the average from the final exam (25%) and the three highest test scores (25% each; out of four). No make up tests will be given. If you miss a test for any reason that is the one you will drop. No extra time will be given if you arrive late for a test, so please arrive on time.

**You must take the final exam to pass the class.**

Note: Extra credit is not allowed for this class, and work competed for another class will not be accepted in this class.
**Attendance and Rules:** Please make sure that you sign the attendance sheet during each class. If you arrive late, please make sure you remember to sign at the end of class. If there is more than one unexcused absence before each test, I reserve the right to drop you from the class.

Unsocial behavior such as talking during class or being disruptive will not be tolerated and you will be asked to leave the class. Please remember to silence your cell phones during class.

All communications outside of class should be via campus mail (myUWG), and you are expected to check your mail regularly.

**Suggested problems:** No homework will be collected. However, it is expected that you will attempt to solve the in chapter problems including redoing the problems from class. After reading a topic you should attempt to solve the problems relating to that topic at the end of each chapter. If you have time, work on solving some of the “General Problems” at the end of each chapter. You should also attempt the problems from the ACS book (selected problems are posted on CourseDen). Obviously, the more problems you solve the better understanding you will have for the material. The best way to make sure that you have understood the material covered in class is to work through the appropriate problems on a regular basis and not just a couple of days before the tests.

**Exams:** Cell phones, calculators, cameras and all other electronic devices must be stored away and may not be visible during exams. The final is a comprehensive exam over the entire course. I reserve the right to ask you to sit at an assigned seat in the classroom during exam days.

**Scantrons:** You will need a total of 5 large blue scantron forms (#SC984-E) for the tests and final. Folded or crumpled forms will not be accepted. **These forms will NOT be available in class.**

**Academic Honesty:**

- All exams will be closed book/closed notes, and will be taken individually.
- During exams you may not use your own paper or other materials except your pencil and scantron.
- Academic dishonesty will not be tolerated. Academic dishonesty includes unauthorized use of any materials, notes, sources of information, electronic equipment, or study aids during the exam. It also includes the unauthorized assistance of any person other than the course instructor during the exam, the unauthorized viewing of another person’s work during the exam, or the unauthorized securing of all or part of any exam before submission by the instructor.
- Violation of academic honesty will generate disciplinary action that may include a course grade of F. A student who is suspected of cheating must confess to all wrong doing at the first opportunity (when first confronted), or risk a harsher penalty. If you believe that there are situations in the course that foster academic dishonesty, please bring them to my attention. Likewise, if you have observed cheating, bring the details to my attention as soon as practical. Insofar as it is possible, your anonymity will be protected.

**Some Advice:** This class demands your time and attention. If you hope to do well in this class you should at the very least attend class, pay attention during class and be prepared to spend 9 hours per week of quality time reading the book, adding your own notes to the PowerPoint notes provided (on CourseDen), and work as many problems as time allows. You should read over the material to be covered in the next class period ahead of time so that once you are in class you will already be somewhat familiar of the topics covered. I will typically cover approximately 20 pages from the text per class period, so you should read ahead by that amount. It is truly in your best interest to master the material covered in part I, as you will build on this in part II. If you do not do well in part I you will most likely not do well in part II. Self-discipline and developing good study habits are key!

**Office hours:** TR 11 AM – 1 PM & 5 – 6 PM; W 11 AM – 2 PM & 5 – 6 PM
All Students Please Note!

For important policy information, i.e., the UWG Honor Code, Email, Credit Hour, and Campus Carry policies, as well as information on Academic Support and Online Courses, please review the information found in the Common Language for Course Syllabi documentation at https://www.westga.edu/administration/vpaa/assets/docs/faculty-resources/common_language_for_course_syllabi_v2.pdf. Additions and updates are made as institution, state, and federal standards change, so please review it each semester.

Learning Outcomes

1. Reason and think analytically in solving problems and making decisions in matters involving organic chemistry. Attainment of this learning outcome will be reflected by the students ability to:
   - Select reagents to accomplish appropriate functional group transformation.
   - Draw the product of a chemical reaction.
   - Indicate the appropriate stereochemistry of organic molecules.

2. Apply a basic understanding of the systematic methods of scientific inquiry, principles and procedures to investigate problems. Attainment of this learning outcome will be reflected by the students ability to:
   - Describe the mechanism of particular organic reactions.
   - Analyze IR and Mass Spectral, and NMR data to determine the structure of organic molecules.

3. To communicate organic chemistry with clarity.

   Successfully complete written and oral assignments, and examinations.